

Novel Synthesis of Thioguanine and Sulfanylpurine Analogues: Reaction of Heterocyclic Ketene Dithioacetals with Nucleophiles†

Galal H. Elgemeie,^{*a} Ahmed H. Elghandour,^b Ali M. Elzanate^b and Sayed A. Ahmed^c

^aChemistry Department, Faculty of Science, Helwan University, Cairo, Helwan, Egypt

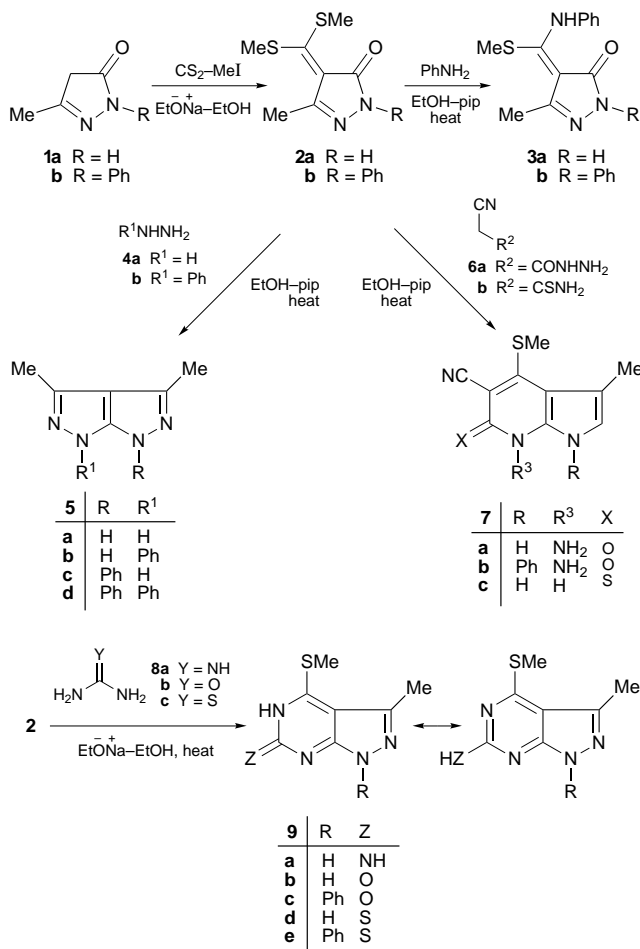
^bChemistry Department, Faculty of Science, Cairo University (Bani Suef Branch), Bani Suef, Egypt

A novel synthesis of thioguanine and sulfanylpurine analogues *via* the reaction of heterocyclic ketene dithioacetals with nucleophiles is reported and the synthetic potential of the method is demonstrated.

Synthetic analogues of purines are widely used in medical science and clinical medicine. The purine analogues of purines are widely used in medical science and clinical medicine. The purine analogue 4-hydroxypyrazolopyrimidine (allo-purinol), used in treatment of hyperuricemia and gout, inhibits *de novo* purine biosynthesis and xanthine oxidase activity. Both 6-thioguanine and 6-sulfanylpurine, in which thiol groups replace the hydroxy groups at the 6-position, are widely used clinically. As part of our programme of research on the synthesis of purine analogues and other antimetabolites^{1–7} we have recently reported an interesting synthesis of 7-methylsulfanylpurazolo[1,5-*a*]pyrimidines and 7-methylsulfanylpurazolo[1,2-*c*]imidazolones *via* the reaction of cyanoketene dithioacetals with 5-aminopyrazoles and 2-sulfanylhantoinins, respectively.^{8–10} We report here a novel and convenient method for the synthesis of fused pyrazoles carrying a methylsulfanyl group. Derivatives of these ring systems are interesting because they are sulfanylpurine analogues and as such they may have useful properties as antimetabolites in purine biochemical reactions. The heterocyclic ketene dithioacetals **2** were chosen as the key intermediate and were prepared by the reaction of pyrazolin-5-ones **1** with sodium ethoxide and carbon disulfide, followed by methyl iodide treatment in a one-pot reaction (Scheme 1). The structure of **2** was established on the basis of elemental analysis and spectral data. Thus, the mass spectrum of **2a** was compatible with the molecular formula C₇H₁₀N₂OS₂ (M⁺ 202) and its ¹H NMR spectrum contained two singlets at δ 2.52 and 2.63 ppm, which are assignable to two methylsulfanyl groups, and a broad band at 12.22 ppm assignable to an NH group.

The ¹³C NMR spectrum was characterized by two signals, at 18.20 and 18.70 ppm corresponding to two SMe carbons. Compounds **2** reacted with aromatic amines in refluxing ethanol containing catalytic amounts of piperidine to afford the corresponding anilino derivatives **3**. The structures of compound **3** were established on the basis of elemental analysis and spectral data. Thus, the mass spectrum of **3a** was compatible with the molecular formula C₁₂H₁₃N₃OS (M⁺ 247) and its ¹H NMR spectrum contained a band at 2.52 ppm, assigned to the SMe group, and two broad bands at 10.17 and 11.00 ppm assignable to two NH groups. Compounds **2** bearing latent functional substituents were found useful for the synthesis of fused pyrazole derivatives. Thus, it has been found that the reaction of ketene dithioacetals **2** with hydrazine hydrate in refluxing ethanol containing catalytic amounts of piperidine gives the corresponding 4-methylsulfanylpurazolo[3,4-*c*]pyrazoles (**5**)

in good yields. The structures of **5** were established and confirmed for the reaction products on the basis of their elemental analysis and spectral data (MS, IR and ¹H NMR). The analytical data for **5a** indicated a molecular formula C₆H₈N₄S (M⁺ 168) and ¹H NMR revealed a band at δ 2.74 ppm, assignable to a SMe group. When compounds **2** were subjected to reaction with cyanoacetohydrazide **6a** or cyanothioacetamide **6b** in refluxing ethanol containing catalytic amounts of piperidine, the corresponding 4-methylsulfanylpurazolo[3,4-*b*]pyridine derivatives **7** were obtained and their structures were established on the basis of elemental analysis and spectral data. Thus, for **7b** the IR spectrum revealed the presence of a cyano group at 2220 cm⁻¹, and the ¹H NMR spectrum revealed a band at δ 2.56 ppm, assignable to the SMe group, a multiplet at δ 6.91–8.70 ppm, assigned for aromatic protons, and a broad singlet at δ 3.35 ppm assigned to the amino group. The formation



Scheme 1

*To receive any correspondence.

†This is a **Short Paper** as defined in the Instructions for Authors, Section 5.0 [see *J. Chem. Research (S)*, 1998, Issue 1]; there is therefore no corresponding material in *J. Chem. Research (M)*.

of **7** from the reaction of dithioacetals **2** and cyanoacetohydrazide **6a** is assumed to proceed *via* the intermediacy of Michael adducts, which cyclize to yield the final 4-methylsulfanylpyrazolo[3,4-*b*]pyridin-6-one **7**. Direct treatment of **2** with guanidine, urea or thiourea afforded the corresponding 4-methylsulfanylpyrazolo[3,4-*d*]pyrimidine derivatives **9**. The structures of compounds **9** were established on the basis of their elemental analysis and spectral data. The ¹H NMR spectrum of **9a** showed a broad band at δ 6.80 ppm assignable to an amino group. The mass spectrum was compatible with the molecular formula C₇H₉N₅S (M⁺ 196).

In summary, we have achieved a regioselective synthesis of interesting non-classical sulfanylpyrimidine and thioguanine analogues and other antimetabolites by the reaction of heterocyclic ketone dithioacetals with hydrazine derivatives and active methylene compounds. The compounds obtained seem promising for further chemical transformations and biological evaluation studies.

Experimental

All mps are uncorrected. The IR spectra were obtained (KBr disk) on a Perkin Elmer/1650 FT-IR instrument. The ¹H NMR spectra were measured on a Varian 400 MHz spectrometer for solutions in (CD₃)₂SO using SiMe₄ as an internal standard. Mass spectra were recorded on a Varian MAT 112 spectrometer. Analytical data were obtained from the Microanalytical Data Center at Cairo University.

4-Bis(methylsulfanyl)methylidene-3-methylpyrazolin-5-ones (2a,b). *General procedure.*—A mixture of 3-methylpyrazolin-5-ones (**1**) (0.01 mol) and a solution of sodium ethoxide (0.02 mol) was boiled under reflux for 20 min, then cooled and carbon disulfide (0.01 mol) was added. The reaction mixture was warmed at 30 °C for 20 min. After cooling, methyl iodide (0.02 mol) was added. The mixture was poured over an ice-water mixture and neutralized with dil. hydrochloric acid. The precipitated product was collected by filtration and recrystallized from ethanol **2a**. Yellow crystals, from EtOH, yield 95%, mp 202 °C; $\nu_{\max}/\text{cm}^{-1}$ (KBr) 1655 (CO), 1525 (C=N); δ_{H} [(CD₃)₂SO] 2.51 (s, 3 H, CH₃), 2.52 (s, 3 H, SCH₃), 2.63 (s, 3 H, SCH₃), 12.22 (s, br, 1 H, NH), δ_{C} 11.51 (CH₃), 18.22 (SCH₃), 18.70 (SCH₃), 113.10 (C-6), 143.00 (C-4), 160.00 (C-3), 183.00 (C-5); *m/z* 202 (Found: C, 41.8; H, 4.5; N, 13.6. C₇H₁₀N₂O₂S requires C, 41.6; H, 4.9; N, 13.8%). **2b**: Yellow crystals, from EtOH, yield 95%, mp 94 °C; $\nu_{\max}/\text{cm}^{-1}$ (KBr) 1655 (CO), 1525 (C=N); *m/z* 278 (Found: C, 56.4; H, 5.1; N, 10.3. C₇H₁₀N₂O₂S₂ requires C, 56.1; H, 5.0; N, 10.0%).

4-Substituted-3-methylpyrazolin-5-one derivatives (3a,b). *General procedure.*—A mixture of pyrazolinones **2a,b** (0.01 mol) and aniline (0.01 mol) was boiled under reflux in ethanol (30 ml), containing a catalytic amount of piperidine for 3 h. The product was isolated after cooling, and recrystallized from ethanol. **3a**: Yellow crystals, from EtOH, yield 80%, mp 235 °C; $\nu_{\max}/\text{cm}^{-1}$ (KBr) 3426 (NH), 1657 (CO), δ_{H} [(CD₃)₂SO] 2.40 (s, 3 H, CH₃), 2.52 (s, 3 H, SCH₃); 7.1–7.8 (m, 5 H, C₆H₅), 10.17 (s, br, 1 H, NH), 11.00 (s, br, 1 H, NH); *m/z* 247 (Found: C, 58.6; H, 4.9; N, 17.2. C₁₂H₁₃N₃O₂S requires C, 58.3; H, 5.2; N, 17.0). **3b**: Yellow crystals, from EtOH, yield 55%, mp > 300 °C; $\nu_{\max}/\text{cm}^{-1}$ (KBr) 3178 (NH), 1668 (CO), 1536 (C=N) (Found: C, 66.5; H, 5.1; N, 13.4. C₁₈H₁₇N₃O₂S requires C, 66.8; H, 5.3; N, 13.0%).

3-Methyl-4-methylsulfanylpyrazolo[4,3-*c*]pyrazoles (3a,b). *General procedure.*—A solution of pyrazolinones **2a,b** (0.01 mol) and hydrazine hydrate or phenylhydrazine (0.01 mol) in ethanol (30 ml) containing a catalytic amount of piperidine was boiled under reflux for 3 h. The solution mixture was left to cool, and the product which separated was collected and recrystallized from the appropriate solvent. **5a**: Buff crystals, from DMF, yield 75%, mp > 300 °C; $\nu_{\max}/\text{cm}^{-1}$ (KBr) 3341, 3310, 3270 (NH), 1538 (C=N); δ_{H} [(CD₃)₂SO] 2.51 (s, 3 H, CH₃), 2.74 (s, 3 H, SCH₃), 7.96 (s, br, 1 H, NH), 9.78 (s, br, 1 H, NH); *m/z* 168 (Found: C, 42.5; H, 4.5; N, 33.6. C₆H₈N₄S requires C, 42.8; H, 4.7; N, 33.3%). **5b**: Green crystals, from EtOH, yield 60%, mp 190 °C; $\nu_{\max}/\text{cm}^{-1}$ (KBr) 3384, 3182 (NH), 1623 (C=N); (Found: C, 59.4; H, 5.1; N, 23.2. C₁₂H₁₂N₄S₂ requires C, 59.0; H, 4.9; N, 22.9%). **5c**: Buff crystals, from DMF, yield 60%, mp > 300 °C; $\nu_{\max}/\text{cm}^{-1}$ (KBr) 3404, 3240,

3035 (NH), 1594 (C=N); δ_{H} [(CD₃)₂SO] 2.36 (s, 3 H, CH₃), 2.53 (s, 3 H, SCH₃), 7.08–8.00 (m, 5 H, C₆H₅), 9.52 (s, br, 1 H, NH) (Found: C, 59.2; H, 5.2; N, 22.7. C₁₂H₁₂N₄S₂ requires C, 59.0; H, 4.9; N, 22.9%). **5d**: Black crystals, from EtOH, yield 75%, mp 203 °C; $\nu_{\max}/\text{cm}^{-1}$ (KBr) 1628 (C=N) (Found: C, 67.1; H, 4.8; N, 17.3. C₁₈H₁₆N₄S₂ requires C, 67.5; H, 5.0; N, 17.5%).

5-Cyano-3-methyl-4-methylsulfanylpyrazolo[3,4-*b*]pyridines (7a-c).—A mixture of pyrazolinones **2** (0.01 mol) and cyanoacetohydrazide or cyanothioacetamide (0.01 mol) was boiled under reflux in ethanol (20 ml) containing a catalytic amount of piperidine for 6 h. The product was collected and recrystallized from the appropriate solvent. **7a**: Yellow crystals, from EtOH–DMF, yield 55% mp > 300 °C; $\nu_{\max}/\text{cm}^{-1}$ (KBr) 3300, 3230 (NH₂, NH), 2201 (CN), 1660 (CO), 1583 (C=N); *m/z* 235 (Found: C, 45.5; H, 4.1; N, 29.5. C₉H₉N₅O₂S requires C, 45.9; H, 3.8; N, 29.8%). **7b**: Yellow crystals, from EtOH, yield 55%, mp 165 °C; $\nu_{\max}/\text{cm}^{-1}$ (KBr) 3204, 3008 (NH₂, NH), 2220 (CN), 1726 (CO), 1569 (C=N); δ_{H} [(CD₃)₂SO] 2.40 (s, 3 H, CH₃), 2.56 (s, 3 H, SCH₃), 3.35 (s, 2 H, N–NH₂), 6.91–8.70 (m, 5 H, C₆H₅) (Found: C, 57.6; H, 4.2; N, 22.8. C₁₅H₁₃N₅O₂S requires C, 57.9; H, 4.2; N, 22.5%). **7c**: Brown crystals, from EtOH–DMF, yield 50%, mp > 300 °C; $\nu_{\max}/\text{cm}^{-1}$ (KBr) 3328, 3140 (NH), 2201 (CN); *m/z* 236 (Found: C, 45.4; H, 3.6; N, 23.3. C₉H₈N₄S₂ requires C, 45.8; H, 3.4; N, 23.7%).

3-Methyl-4-methylsulfanylpyrazolo[3,4-*d*]pyrimidines (9a-e).—*Method (A)*: Guanidine hydrochloride (0.01 mol) was heated in sodium ethoxide (0.01 mol) in EtOH (30 ml) for 30 min, then an equivalent amount of pyrazolinone **2a** (0.01 mol) was added. The reaction mixture was boiled under reflux for 3 h. The product was obtained after treatment with dil. HCl and recrystallized from ethanol. **9a**: Yellow crystals from EtOH, yield 55%, mp > 300 °C; $\nu_{\max}/\text{cm}^{-1}$ (KBr) 3637, 3443, 3302 (NH₂, NH), 1553 (C=N); δ_{H} [(CD₃)₂SO] 2.38 (s, 3 H, CH₃), 2.51 (s, 3 H, SCH₃), 6.80 (s, 2 H, NH₂), 10.85 (s, br, 1 H, NH); *m/z* 196 (Found: C, 43.4; H, 4.5; N, 35.5. C₇H₉N₅S requires C, 43.0; H, 4.1; N, 35.9%). *Method (B)*: Equivalent amounts of pyrazolinones **2a,b** (0.01 mol) and thiourea or urea (0.01 mol) were mixed together and heated to 190 °C for 30 min. The solidified product was collected and recrystallized from ethanol. **9b**: Yellow crystals, from EtOH, yield 55%, mp > 300 °C; $\nu_{\max}/\text{cm}^{-1}$ (KBr) 3493, 3181 (NH), 1700 (CO) (Found: C, 42.4; H, 4.4; N, 28.1. C₇H₈N₄O₂S requires C, 42.8; H, 4.0; N, 28.5%). **9c**: Yellow crystals, from EtOH, yield 60%, mp > 300 °C; $\nu_{\max}/\text{cm}^{-1}$ (KBr) 3366 (NH), 1734 (CO), 1561 (C=N) (Found: C, 57.1; H, 4.6; N, 20.2. C₁₃H₁₂N₄O₂S requires C, 57.3; H, 4.4; N, 20.5%). **9d**: Yellow crystals, from EtOH, yield 60%, mp > 300 °C; $\nu_{\max}/\text{cm}^{-1}$ (KBr) 3708, 3356 (NH), 1556 (C=N) (Found: C, 39.1; H, 4.2; N, 26.2. C₇H₈N₄S₂ requires C, 39.6; H, 3.7; N, 26.4%). **9e**: Yellow crystals, from EtOH yield 55%, mp > 300 °C; $\nu_{\max}/\text{cm}^{-1}$ (KBr) 3100 (NH), 1593 (C=N); δ_{H} [(CD₃)₂SO] 2.31 (s, 3 H, CH₃); 2.65 (s, 3 H, SCH₃); 7.38–7.71 (m, 5 H, C₆H₅) (Found: C, 54.4; H, 4.4; N, 19.1. C₁₃H₁₂N₄S₂ requires C, 54.1; H, 4.2; N, 19.4%).

Received, 1st July 1997; Accepted, 22nd October 1997
Paper E/7/04612J

References

- G. E. H. Elgemeie, A. M. Attia, A. M. Elzanaty and A. K. Mansour, *Bull. Chem. Soc. Jn.*, 1994, **67**, 1627.
- G. E. H. Elgemeie, A. M. Attia, H. A. Ali and A. K. Mansour, *J. Chem. Res. (S)*, 1994, **78**.
- G. E. H. Elgemeie, A. M. Attia, D. S. Farag and S. M. Sherif, *J. Chem. Soc., Perkin Trans. 1*, 1994, 285.
- G. E. H. Elgemeie and B. A. W. Hussain, *Tetrahedron*, 1994, **50**, 199.
- G. E. H. Elgemeie, A. M. Attia and N. M. Fathy, *Liebigs Ann. Chem.*, 1994, 955.
- G. E. H. Elgemeie and A. M. Attia, *Carbohydr. Res.*, 1995, 268, 295.
- G. E. H. Elgemeie and N. M. Fathy, *Tetrahedron*, 1995, **51**, 3345.
- G. E. H. Elgemeie, S. E. Elezbawy, H. A. Ali and A. K. Mansour, *Bull. Chem. Soc. Jpn.*, 1994, **67**, 738.
- G. E. H. Elgemeie, H. A. Ali and A. K. Mansour, *Phosphorus, Sulfur Silicon*, 1994, **90**, 143.
- G. E. H. Elgemeie, H. A. Ali and A. M. Elzanaty, *J. Chem. Res. (S)*, 1996, 340.